The contribution of limiting nutrients to the ocean from the 1991 eruption of Mount Pinatubo

Jui-Chen Chien^{1,2}, and Tung-Yuan Ho¹

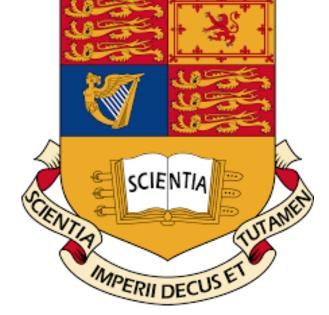
¹Research Center for Environmental Change, Academia Sinica, Taipei, Taiwan

²Department of Earth Science and Engineering, Imperial College London, South Kensington, London SW7 2AZ, UK

Background and Motivation

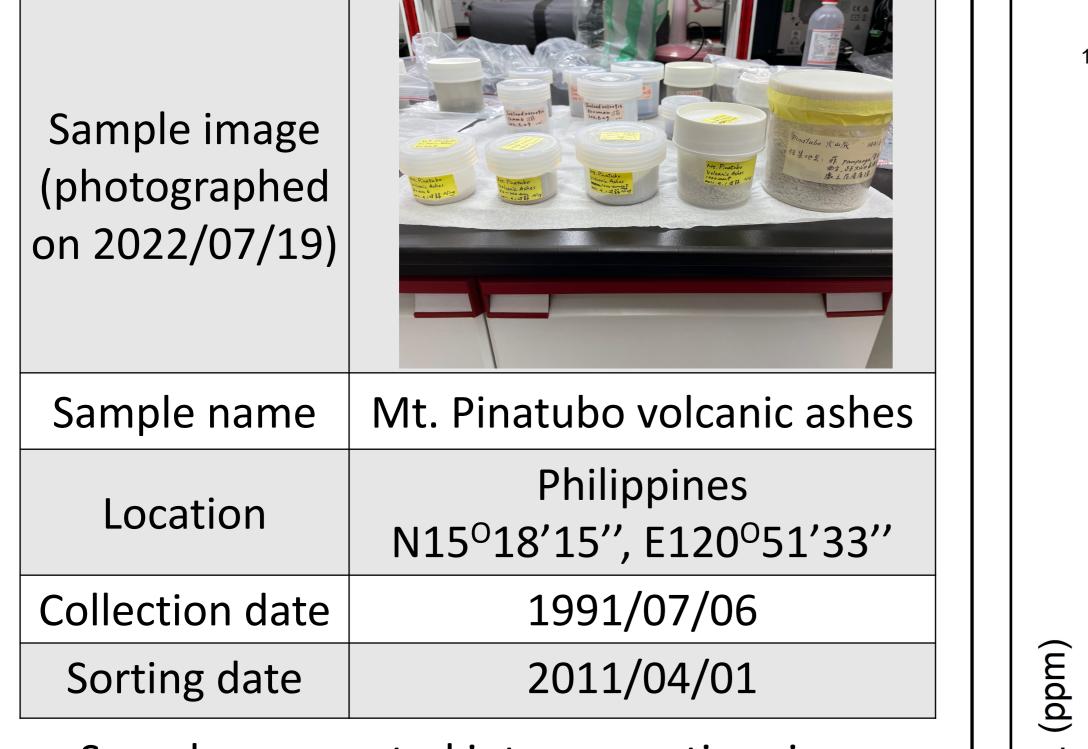
The 1991 eruption of Mount Pinatubo was the second-largest volcanic event in the 20th century, producing a bulk tephra volume of 5.5km³, with a Volcanic Explosivity Index (VEI) of $6^{(1)}$. This eruption lowered the global CO₂ concentration by 2 ppm! It was proposed to be attributed to the effect of surface ocean fertilization from the release of limiting nutrients from the tephra⁽²⁾.

Previous studies suggested that volcanic ashes could enhance biological fixation of atmospheric $CO_2^{(3)}$, due to the significant input of bioavailable major and minor nutrients^(4,5,6). However, to the best of our knowledge, no study has reported and estimated the contribution of the limiting

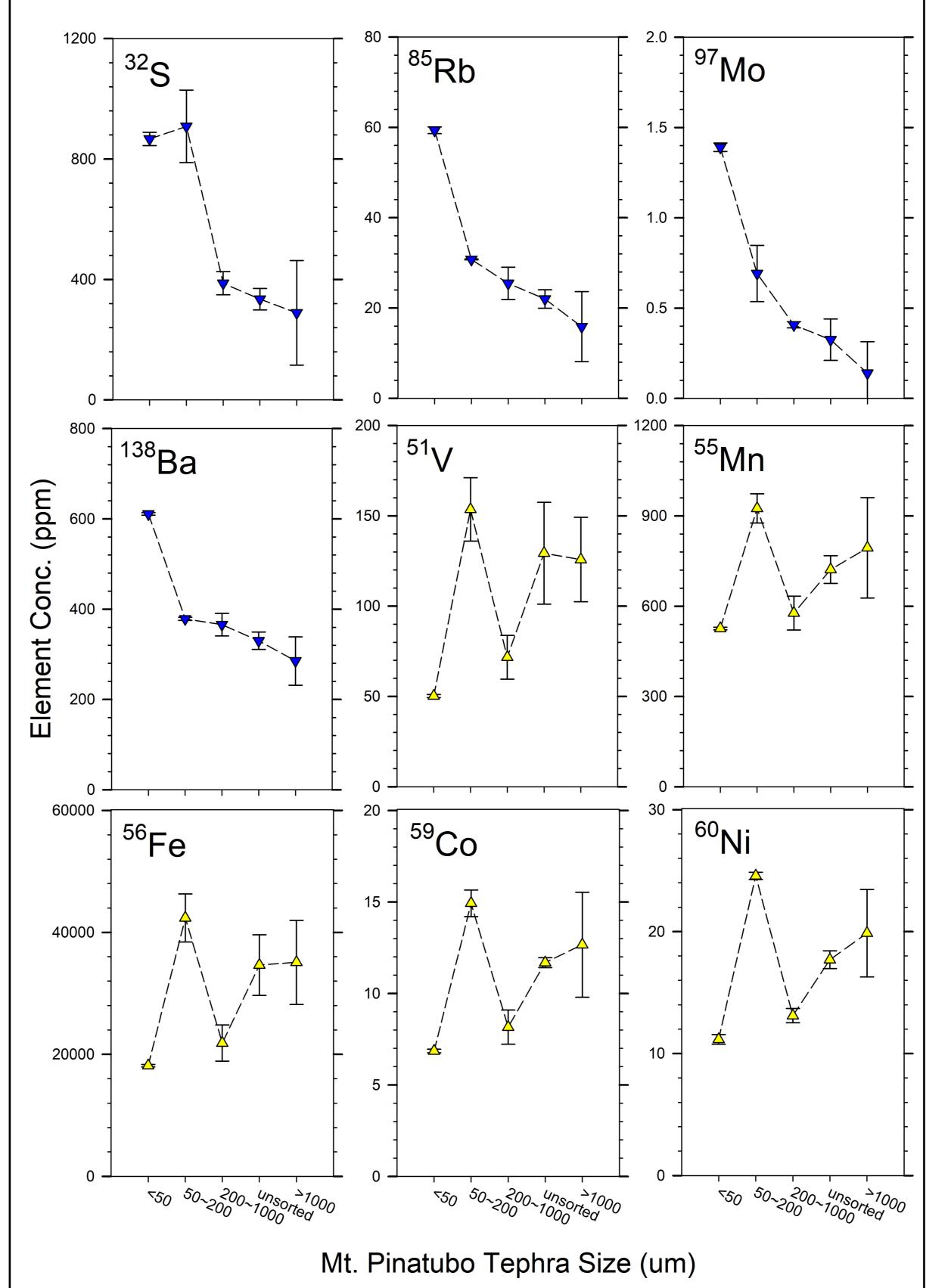


nutrients to the ocean, particularly P and Fe, from the 1991 eruption. Therefore, this study attempts to quantify Mt. Pinatubo's limiting nutrient contribution.

Samples



Samples are sorted into respective sizes. <50um, 50-200um, 200-1000um, >1000um



Results and Discussion

Shown Fig. 1, the elements in blue exhibits a negative trend, proposed to be the effect of adsorption onto tephra favoring small size (high surface area to volume ratio). S, Rb, and Ba's low melting point to volcanic eruptions (up to 1200 ^oC) results in readily adsorption. The reason behind a negative trend of Mo is uncertain. Fe and P, two limiting nutrients, are critical factors influencing phytoplankton growth in the ocean. The estimated contribution of

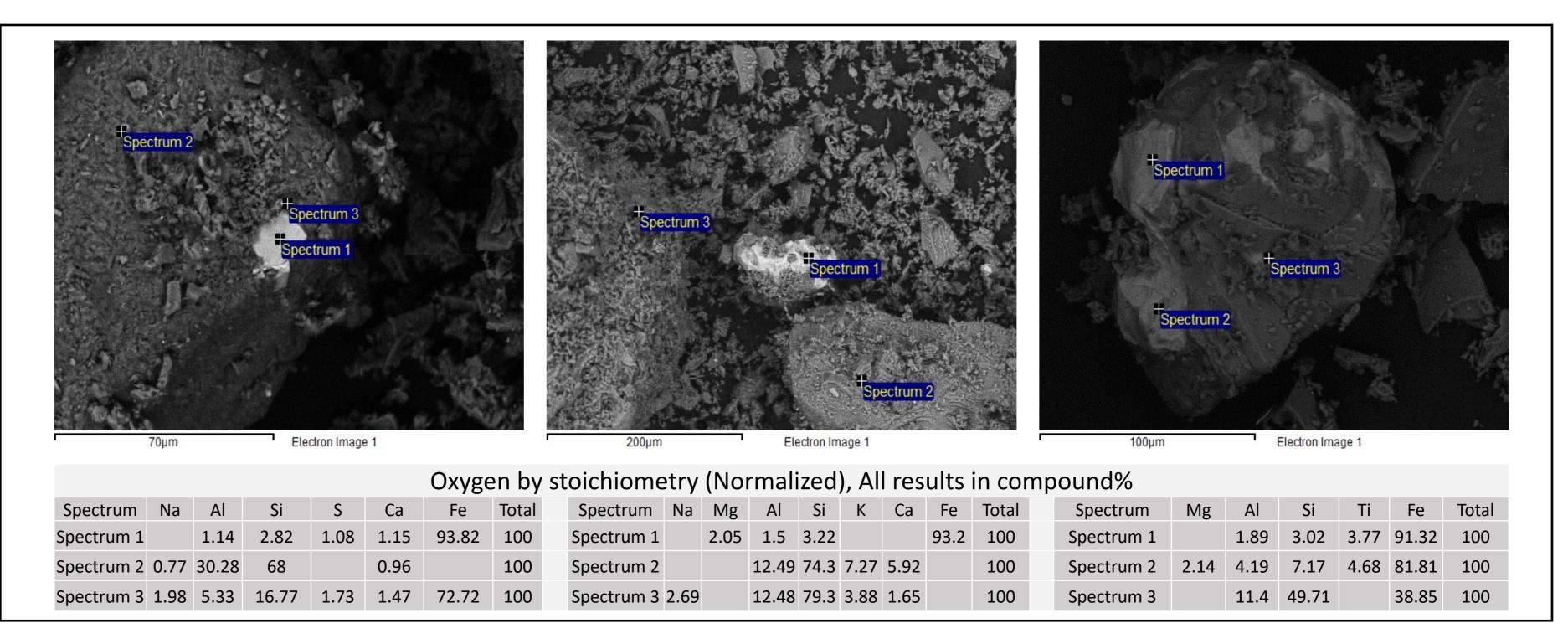
dissolvable Fe and P from the eruption are 10¹⁰ and 10⁹ mole, respectively, which theoretically would result in 10¹⁵ and 10¹¹ mole of CO₂ uptake, respectively.

Method

- The samples were collected by Prof. Shu-Chen Pai of Institute of Oceanography, National Taiwan University
- Sample sizes are validated using Scanning Electron Microscope (SEM)
- Energy Dispersive Spectroscopy (EDS) is used for mineralogical analysis of samples.
- Sample's bulk element concentration is obtained by Inductively Coupled Plasma Mass Spectrometry (ICP-MS) using total digestion procedure $^{(7)}$.
- To simulate tephra's instant dissolution in seawater, ultrapure water leaching

Figure 1. A Selection of Bulk Element Concentration in Different Size Fractions

Fig. 2 shows increased occurrence of iron oxide in size 50-200um, supporting ⁵⁶Fe data in Fig. 1. This could be explained by crystal's preferred size stabilization. Hematite (Fe₂O₃), Ilmenite (FeTiO₃) and Pyrite (FeS₂) are also found in greater abundance in size 50-200um.



method is used⁽⁷⁾.

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Figure 2. Mineral composition analysis on size 50~200um using SEM and EDS

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